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Chapter 7 Review Chemical Formulas

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

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7 Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: a. copper(II) carbonate b. sodium sulfite c. ammonium phosphate d. tin(IV) sulfide e. nitrous acid 2. Write the Stock system names for the following compounds: a. $\text{Mg}(\text{ClO}_4)_2$ b. $\text{Fe}(\text{NO}_3)_2$

7 Chemical Formulas and Chemical Compounds

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided 1 c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit (b) how many sulfate ions can be attached to the iron atom (c)

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7 is the final chapter of the first semester. It serves as a summary of all course content discussed so far, as the focus of this chapter...

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following

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questions in the space provided. 1. Assign the oxidation number to the specified element in each of the following examples: a. S in H_2SO_3 b. S in MgSO_4 c. S in K_2S d. Cu in Cu_2S e. Cr in Na_2CrO_4 f. N in HNO_3 g. C in (HCO_3) h. N in (NH_4) 2. a.

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Chapter 7 Significance of a Chemical Formula •A chemical formula indicates the relative number of atoms of each kind in a chemical compound. •For a molecular compound, the chemical formula reveals the number of atoms of each element contained in a single molecule of the compound. •example: octane —C₈H₁₈

Chemical Formulas and Chemical Chapter 7 Compounds

A chemical formula includes the symbols of the elements in the compound and subscripts that indicate Chemistry Ch. 7 Review--Chemical Formulas and Chemical Compounds DRAFT 10th - 12th grade

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 2 SHORT ANSWER Answer the following questions in the space provided 1 Assign the oxidation number to the specified element in each of the following examples: 4 a S in H_2SO_3 6 b S in MgSO_4 2 c S in K_2S 1 d Cu in Cu_2S 6 e Cr in Na_2CrO_4 5 f N in HNO_3 4 g C in HCO_3^- [MOBI] Chapter 7 Modern Chemistry Review Answers

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CHAPTER 7 REVIEW. Chemical Formulas and Chemical Compounds. MIXED REVIEW. SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: a.copper(II) carbonate. b.sodium sulfite. c.ammonium phosphate.

CHAPTER 7 REVIEW

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

Modern Chemistry Chapter 7 Review Chemical Formulas And ...

chemical formulas. In this chapter, you will be introduced to some of the rules used to identify simple chemical compounds. Significance of a Chemical Formula Recall that a chemical formula indicates the relative number of atoms of each kind in a chemical compound. For a molecular compound, the

CHAPTER 7 Chemical Formulas and Chemical Compounds

Using Chemical Formulas SECTION 4 Determining Chemical

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Formulas CHAPTER 7 Untitled-2 206 6/28/2011 2:09:06 PM Lab Preview Chapter Overview Section 1 describes the naming of binary ionic and molecular compounds. Section 2 describes the way oxidation numbers are assigned and the Stock system of naming compounds. Section 3 describes how to calculate

1 Core Instruction

CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER 1. In a Stock system

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name such as iron (III) sulfate, the Roman numeral tells us (a) (b) (c) (d) 2.

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